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Key Concepts: Terms in this set (20)

Heat. the transfer of thermal energy from one object to another because of a difference in temperature. Temperature. a measure of how hot or cold an ...

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Summary 16.1 Thermal Energy and Matter Heat flows spontaneously from hot objects to cold objects. • Heat is the transfer of thermal energy from one object to another because of a temperature difference. Temperature is related to the average kinetic energy of the particles in

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Chapter 16 Thermal Energy and Heat

Physical Science Chapter 16: Thermal Energy and Heat. Heat is the transfer of thermal energy from one object to another as the result of a difference in _____ . _____ produces heat.

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Physical Science Chapter 16: Thermal Energy and Heat ...

It states that thermal energy can flow from colder objects to hotter objects only if work is done on the system. Third law of thermodynamics It states that absolute zero cannot be reached.

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is a measure of how hot or cold an

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object is compared to a ref.... Increase in volume of material when its temperature increases. Conduction. is the transfer of thermal energy through touching with no ove.... Heat Engine. is any device that converts heat into work. 36 Terms. etakp. Chapter 16 Thermal Energy and Heat.

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16.1 – Thermal Energy and Matter . Work
and Heat Heat is the transfer of thermal
energy from one object to another
because of temperature differences Heat
flows spontaneously from hot objects to
cold objects Imagine two glasses with

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differing amounts of water in them at the same temperature The glass with ...

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about convection, conduction and
radiation as well as the use of insulation.
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Chapter 16 - Thermal Energy

16.1 Thermal Energy & Matter. Work and Heat. Heat -the transfer of thermal energy from one object to another because of a temperature difference Heat flows from higher temps to lower temps. Temperature is related to the kinetic energy of the particles: particles

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move around as they heat

Chapter 16

Chapter 16: Thermal Energy and Heat.
Tools. Copy this to my account; E-mail to
a friend; Find other activities; Start over;
Help; A B; heat: the transfer of thermal
energy from one object to another
because of a difference in temperature:

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temperature: a measure of how hot or cold an object is compared to a reference point:

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Chapter 16 Thermal Energy and Heat
Section 161 Thermal Energy and Matter
(pages 474-478) This section defines

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heat and describes how work, temperature, and thermal energy are related to heat Thermal expansion and contraction of materials is discussed, and uses of a calorimeter are explained Reading Strategy (page 474)

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heat is required to raise the temperature of a gold earring from 25.00C to 30.00C? The earring weighs 25 grams, and the specific heat of gold is 0.128 J/ge0C. 1. 2. 3. Read and Understand What information are you given? Specific heat = $c = 0.128 \text{ J/g} \cdot \text{OC}$ Mass = $m = 25.0$ grams

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Morgan A. • 33 cards. Heat. the transfer
of thermal energy from one object to
another as the result of a difference in
temperature. True. T/F: On the Celsius
Scale, the reference points for
temperature are the freezing and boiling
points of water. thermal energy ...

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